

Reaction of Sulfene with Heterocyclic *N,N*-Disubstituted
 α -Aminomethyleneketones. XI. Synthesis of
 1,2-Oxathiino[5,6-*d*]-1-benzoxepin Derivatives
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The 1,4-cycloaddition of sulfene to *N,N*-disubstituted (*E*)-4-aminomethylene-3,4-dihydro-1-benzoxepin-5(2*H*)-ones gave, generally in excellent yield, *N,N*-disubstituted 4-amino-3,4,5,6-tetrahydro-1,2-oxathiino[5,6-*d*]-1-benzoxepin 2,2-dioxides, which are derivatives of the new heterocyclic system 1,2-oxathiino[5,6-*d*]-1-benzoxepin. This reaction did not occur only with the *N,N*-diphenylenaminone.

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In pursuing our study of sulfene 1,4-cycloaddition to heterocyclic *N,N*-disubstituted α -aminomethyleneketones in order to synthesize new heterocyclic systems derived from 1,2-oxathiin and incorporating potential pharmacologically active molecules (1), we now wish to report the reaction of *N,N*-disubstituted (*E*)-4-aminomethylene-3,4-dihydro-1-benzoxepin-5(2*H*)-ones III with sulfene to afford derivatives of a new heterocyclic system incorporating the pharmacologically interesting 1-benzoxepin moiety (2), namely 1,2-oxathiino[5,6-*d*]-1-benzoxepin (3).

The starting enaminones IIIa-h (Table I) were prepared in excellent yield from 3,4-dihydro-4-hydroxymethylene-1-benzoxepin-5(2*H*)-one II and secondary amines, following previously described procedures (6,7). They are probably *E* isomers, at least as can be seen from the upfield shifts of the CH₂-2 (~0.15-0.3 δ) and CH₂-3 (~0.35-0.6 δ) protons caused by the phenyl group(s) in compounds

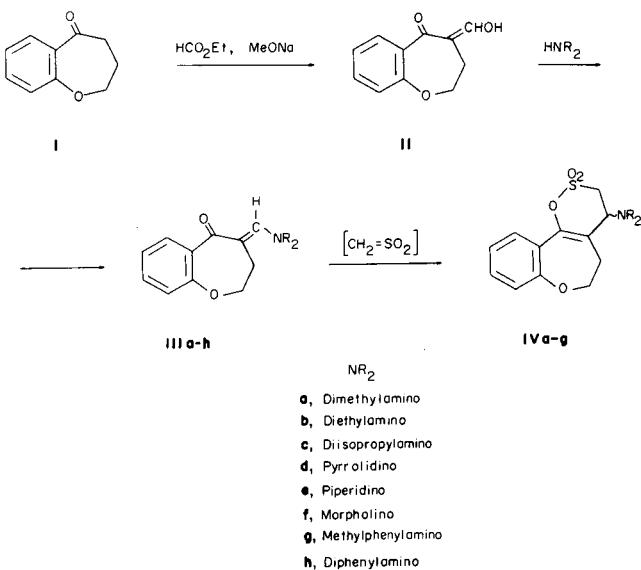


Table I

N,N-Disubstituted (*E*)-4-Aminomethylene-3,4-dihydro-1-benzoxepin-5(2*H*)-ones (a)

Formula Number	NR ₂	Yield %	Mp °C (b)	Molecular Formula	Analyses %		
					Calcd.	Found	N
IIIa	Dimethylamino	91	104	C ₁₃ H ₁₅ NO ₂	71.87 71.64	6.96 7.07	6.45 6.36
IIIb	Diethylamino	77	65	C ₁₅ H ₁₉ NO ₂	73.44 73.71	7.81 7.81	5.71 5.63
IIIc	Diisopropylamino	83	116	C ₁₇ H ₂₃ NO ₂	74.97 75.20	8.39 8.64	5.07 5.22
IIId	Pyrrolidino	97	107	C ₁₅ H ₁₇ NO ₂	74.05 73.75	7.04 6.85	5.76 5.76
IIIe	Piperidino	96	105	C ₁₆ H ₁₉ NO ₂	74.68 74.40	7.44 7.62	5.44 5.47
IIIf	Morpholino	96	150	C ₁₅ H ₁₇ NO ₃	69.48 69.30	6.61 6.50	5.40 5.36
II Ig	Methylphenylamino	89	89	C ₁₈ H ₁₇ NO ₂	77.40 77.42	6.13 6.26	5.01 5.00
II Ih	Diphenylamino	82	134	C ₂₃ H ₁₉ NO ₂	80.92 80.81	5.61 5.75	4.10 3.99

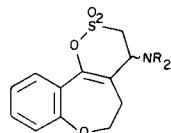
(a) Enaminones IIa,b,d,e,f and IIc,g,h were prepared according to the literature (6) and (7), respectively. (b) From ethyl acetate.

Table II
UV, IR and NMR Spectral Data of Compounds IIIa-h

	UV λ max nm (log ϵ)	C=O	IR, cm ⁻¹ C=C	NMR, δ
IIIa	214 (4.08) 250.5 (3.89) 351 (4.24)	1652	1545	2.74 (near t, J = 6, CH ₂ -3), 3.12 (s, 2 CH ₃ N), 4.26 (t, J = 6, CH ₂ -2), 6.90-7.55 (m, 3 H aryl), 7.73 (dd, J' = 7.2, J'' = 2.4, CH-6), 7.74 (near s, =CHN)
IIIb	214 (4.01) 250.5 (3.80) 351 (4.18)	1646	1535	1.24 (t, J = 7.2, 2 CH ₃), 2.67 (t, J = 6, CH ₂ -3), 3.36 (q, J = 7.2, 2 CH ₂ N), 4.27 (t, J = 6, CH ₂ -2), 6.9-7.9 (m, 4 H aryl), 7.82 (mc, =CHN)
IIIc	211 (4.10) 250.5 (3.90) 355.5 (4.31)	1640	1523	1.25 (d, J = 6.6, 4 CH ₃), 2.67 (t, J = 6, CH ₂ -3), 3.86 (h, J = 6.6, 2 CHN), 4.24 (t, J = 6, CH ₂ -2), 6.90-7.55 (m, 3 H aryl), 7.77 (dd, J' = 7.2, J'' = 2.4, CH-6), 8.01 (near s, =CHN)
IIId	212.5 (3.95) 252.5 (3.77) 355.5 (4.24)	1645	1530	1.91 (mc, 2 CH ₂ pyrr), 2.70 (t, J = 6, CH ₂ -3), 3.54 (mc, 2 CH ₂ N), 4.23 (t, J = 6, CH ₂ -2), 6.9-7.6 (m, 3 H aryl), 7.75 (dd, J' = 7.2, J'' = 2.4, CH-6), 7.97 (near s, =CHN)
IIIE	213.5 (4.03) 251.5 (3.85) 355 (4.27)	1643	1530	1.64 (mc, 3 CH ₂ pip), 2.68 (t, J = 5.4, CH ₂ -3), 3.42 (mc, 2 CH ₂ N), 4.23 (t, J = 6, CH ₂ -2), 6.9-7.6 (m, 3 H aryl), 7.83 (near s, =CHN), 7.85 (dd, J' = 7.2, J'' = 2.4, CH-6)
IIIf	213 (4.08) 253 (3.84) 351.5 (4.17)	1651	1537	2.72 (t, J = 6, CH ₂ -3), 3.49 (mc, 2 CH ₂ N), 3.78 (mc, 2 CH ₂ O), 4.26 (t, J = 6, CH ₂ -2), 6.90-7.65 (m, 3 H aryl), 7.65 (near s, =CHN), 7.80 (dd, J' = 7.2, J'' = 2.4, CH-6)
II Ig	216.5 (4.08) 256 (3.97) 359 (4.21)	1649	1533	2.34 (t, J = 6, CH ₂ -3), 3.50 (s, CH ₃ N), 4.09 (t, J = 6, CH ₂ -2), 6.9-7.6 (m, C ₆ H ₅ N + 3 H aryl), 7.80 (dd, J' = 7.2, J'' = 2.4, CH-6), 7.94 (near s, =CHN)
II Ih	259.5 (4.05) 373 (4.28)	1652	1539	2.08 (t, J = 6, CH ₂ -3), 3.92 (t, J = 6, CH ₂ -2), 6.85-7.65 (m, 2 C ₆ H ₅ N + 3 H aryl), 7.86 (dd, J' = 7.2, J'' = 2.4, CH-6), 8.10 (near s, =CHN)

Table III

N,N-Disubstituted 4-Amino-3,4,5,6-tetrahydro-1,2-oxathiino[5,6-*d*]-1-benzoxepin 2,2-Dioxides IVa-g (a)



Formula Number	NR ₂	Yield %	Mp °C (b)	Molecular Formula	Analyses %		
					Calcd./Found	C	H
IVa	Dimethylamino	82	120	C ₁₄ H ₁₇ NO ₄ S	56.93	5.80	4.74
IVb	Diethylamino	75	102	C ₁₆ H ₂₁ NO ₄ S	56.88	5.94	4.61
IVc	Diisopropylamino	79	122	C ₁₈ H ₂₅ NO ₄ S	59.42	6.55	4.33
IVd	Pyrrolidino	76	124	C ₁₆ H ₁₉ NO ₄ S	59.66	6.62	4.19
IVe	Piperidino	78	199	C ₁₇ H ₂₁ NO ₄ S	61.51	7.17	3.99
IVf	Morpholino	78	167	C ₁₆ H ₁₉ NO ₅ S	61.79	7.35	4.05
IVg	Methylphenylamino	21	144	C ₁₉ H ₁₉ NO ₄ S	59.79	5.96	4.36
					59.88	5.91	4.23
					60.87	6.31	4.18
					60.63	6.42	4.19
					56.96	5.68	4.15
					57.10	5.89	4.03
					63.85	5.36	3.92
					63.93	5.52	4.06

(a) All compounds were prepared according to the literature (6), using anhydrous THF as the solvent. (b) From 95% ethanol.

IIIg,h in comparison with IIIa-f (Table II).

The reaction of III with methanesulfonyl chloride and triethylamine (sulfene prepared *in situ*) occurred in good yield only in the case of aliphatic *N*-substitution to give *N,N*-disubstituted 4-amino-3,4,5,6-tetrahydro-1,2-oxa-

thiino[5,6-*d*]-1-benzoxepin 2,2-dioxides IVa-g (Table III), whose structure was confirmed by ir and nmr spectral data (Table IV).

Enaminone IIIg (NR₂ = methylphenylamino) gave the corresponding adduct IVg in low yield, whereas IIIh (NR₂

Table IV

IR and NMR Spectral Data of Compounds IVa-g

	IR, cm^{-1}		NMR, δ	
	C=C	O=S=O		
IVa	1667	1380	1191	2.35 (s, 2 CH_3N), 2.67 (mc, $\text{CH}_2\text{-}5$), 3.3-3.6
		1175		(m, $\text{CH}_2\text{-}3$), 3.8-4.6 (m, $\text{CH}\text{-}4 + \text{CH}_2\text{-}6$), 6.9-7.4 (m, 3 H aryl), 7.55-7.80 (m, $\text{CH}\text{-}11$)
IVb	1662	1378	1186	1.10 (t, J = 7.2, 2 CH_3), 2.07-3.13 (m, $\text{CH}_2\text{-}5 + 2 \text{CH}_3\text{N}$), 3.3-3.6 (m, $\text{CH}_2\text{-}3$), 3.95-4.55 (m, $\text{CH}\text{-}4 + \text{CH}_2\text{-}6$), 6.95-7.45
		1172		(m, 3 H aryl), 7.55-7.85 (m, $\text{CH}\text{-}11$)
IVc	1649	1370	1191	1.11 (d, J = 6.6, 4 CH_3), 2.4-2.8 (m, $\text{CH}_2\text{-}5$), 3.22 (h, J = 6.6, 2 CHN), 3.45-3.70 (m, $\text{CH}_2\text{-}3$), 4.00-4.55 (m, $\text{CH}\text{-}4 + \text{CH}_2\text{-}6$), 6.90-7.35 (m, 3 H aryl), 7.55-7.80 (m, $\text{CH}\text{-}11$)
		1181		
IVd	1662	1378	1182	1.80 (mc, 2 CH_2 pyrr), 2.69 (mc, 2 $\text{CH}_2\text{N} + \text{CH}_2\text{-}5$), 3.40-3.75 (m, $\text{CH}_2\text{-}3$), 4.0-4.6 (m, $\text{CH}\text{-}4 + \text{CH}_2\text{-}6$), 6.90-7.45 (m, 3 H aryl), 7.55-7.85 (m, $\text{CH}\text{-}11$)
IVe	1662	1376	1182	1.51 (mc, 3 CH_2 pip), 2.0-2.9 (m, 2 $\text{CH}_2\text{N} + \text{CH}_2\text{-}5$), 3.3-3.7 (m, $\text{CH}_2\text{-}3$), 3.75-4.70 (m, $\text{CH}\text{-}4 + \text{CH}_2\text{-}6$), 6.85-7.40 (m, 3 H aryl), 7.5-7.8 (m, $\text{CH}\text{-}11$)
IVf	1663	1379	1187	2.61 (mc, 2 $\text{CH}_2\text{N} + \text{CH}_2\text{-}5$), 3.35-3.90 (m, $\text{CH}_2\text{-}3 + 2 \text{CH}_2\text{O}$), 3.9-4.7 (m, $\text{CH}\text{-}4 + \text{CH}_2\text{-}6$), 6.95-7.55 (m, 3 H aryl), 7.55-7.85 (m, $\text{CH}\text{-}11$)
IVg	1663	1383	1183	2.54 (mc, $\text{CH}_2\text{-}5$), 2.85 (s, CH_3N), 3.35-3.70 (m, $\text{CH}_2\text{-}3$), 3.95-4.65 (m, $\text{CH}_2\text{-}6$), 5.06 (near t, J = 8.4, $\text{CH}\text{-}4$), 6.65-7.55 (m, $\text{C}_6\text{H}_5\text{N} + 3 \text{H}$ aryl), 7.65-7.90 (m, $\text{CH}\text{-}11$)

= diphenylamino) did not react and was recovered unchanged from the reaction mixture, according to a well established trend of this reaction [cf. (8)].

The biological screening, concerning compounds IIIa-f, included herbicide, insecticide, plant health and *in vitro* antimicrobial activity (9). Only compounds IIIa,b showed a weak activity in the herbicidal area; all the others were found to be inactive.

EXPERIMENTAL

The uv spectra were measured in 95% ethanol with a Hitachi-Perkin-Elmer Model EPS-3T spectrophotometer. The ir spectra were taken in chloroform on a Perkin-Elmer Model 398 spectrophotometer; nmr spectra were recorded in deuteriochloroform on a Perkin-Elmer Model R12 instrument (60 MHz, TMS as the internal standard, J in Hz). Melting points were determined with a Fisher-Johns apparatus.

3,4-Dihydro-4-hydroxymethylene-1-benzoxepin-5(2*H*)-one (II).

This compound was prepared from 3,4-dihydro-1-benzoxepin-5(2*H*)-one I (10) (34 mmoles), ethyl formate (51 mmoles) and sodium methoxide (51 mmoles) in benzene, following a previously described procedure (11), yield, 91%; bp 100-110°/0.4 mm; uv: λ max nm ($\log \epsilon$) 217 (4.27), 250.5 (3.98), 302 (4.04); ir (chloroform): ν max 1635, 1608, 1585, 1558 cm^{-1} ; nmr (deuteriochloroform): δ 2.61 (t, J = 5.1, $\text{CH}_2\text{-}3$), 4.36 (t, J = 4.8, $\text{CH}_2\text{-}2$), 6.95-7.65 (m, 3 H aryl), 7.99 (dd, J' = 7.2, J'' ~ 2, $\text{CH}\text{-}6$), 8.38 (near s, =CH-O), 15.5 (broad s, OH; disappears with deuterium oxide).

Anal. Calcd. for $\text{C}_{11}\text{H}_{10}\text{O}_3$: C, 69.46; H, 5.30. Found: C, 69.33; H, 5.33.

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